

PROCESS FOR PRODUCTION OF A CARBOXYLIC ACID/DIOL MIXTURE SUITABLE FOR USE IN POLYESTER PRODUCTION

5

FIELD OF INVENTION

The present invention relates to a process by which a carboxylic acid/diol mixture is obtained from a slurry or cake carboxylic acid product without isolation of a substantially dry carboxylic acid solid. More 10 specifically, the present invention relates to a process by which a terephthalic acid/diol mixture suitable as a starting material for polyester or co-polyester production is obtained from a slurry or cake terephthalic acid product without isolation of a substantially dry terephthalic acid solid.

15

BACKGROUND OF THE INVENTION

Pursuant to the goal of making polyethylene terephthalate (PET) and 20 other polyesters or co-polyesters, a great deal of patent literature is dedicated to describing the processes for preparing a carboxylic acid/diol mixture suitable as starting material. In general, these inventions describe specific mixing schemes with a purified terephthalic acid solid and liquid ethylene glycol. Additionally, there is substantial body of literature devoted 25 to describing the production of a purified terephthalic acid in powder form that is suitable for use in producing PET and other polyesters or co-polyesters.

The objective of this invention is to describe a process by which the carboxylic acid/diol mixture suitable as a starting material for polyester or co-polyester production is obtained from a slurry or cake carboxylic acid product without isolation of a substantially dry carboxylic acid solid. More 5 specifically, the objective of this invention is to describe a process by which a terephthalic acid/diol mixture suitable as a starting material for polyester or co-polyester production is obtained from a slurry or cake terephthalic acid product without isolation of a substantially dry terephthalic acid solid.

Usually, purified terephthalic acid solid is produced in a multi-step 10 process wherein a crude terephthalic acid is produced. Liquid phase oxidation of p-xylene produces crude terephthalic acid. The crude terephthalic acid does not have sufficient quality for direct use as starting material in commercial PET. Instead, the crude terephthalic acid is usually refined to purified terephthalic acid solid.

15 Usually, in terephthalic acid purification processes, the crude terephthalic acid is dissolved in water and hydrogenated for the purpose of converting 4-carboxybenzaldehyde to p-toluid acid, which is a more water-soluble derivative, and for the purpose of converting characteristically yellow compounds to colorless derivatives. Significant 4- 20 carboxybenzaldehyde or p-toluid acid in the final purified terephthalic acid product is particularly detrimental to polymerization processes as each can act as a chain terminator during the condensation reaction between

terephthalic acid and ethylene glycol in the production of PET. Typical purified terephthalic acid contains on a weight basis less than 25 parts per million (ppm) 4-carboxybenzaldehyde and less than 150 ppm p-toluic acid.

A number of other processes have been developed where a
5 terephthalic acid suitable as starting material for commercial PET production is produced without the use of hydrogenation. Typically, terephthalic acid production processes usually involve catalyzed oxidation of p-xylene in an acetic acid solvent followed by filtration and drying of the terephthalic acid.

10 To produce a terephthalic acid/diol mixture acceptable for PET production from a slurry or cake terephthalic acid product poses a substantially different problem than from a dry terephthalic acid powder.

Typically, terephthalic acid (TPA) produced via catalyzed oxidation of p-xylene in an acetic acid solvent produces a slurry or cake terephthalic
15 acid product that contains residual catalyst (e.g cobalt, manganese, and bromine compounds). In a common method of producing a substantially dry TPA solid from a slurry or cake terephthalic acid product, the slurry or cake terephthalic acid product is filtered to separate a substantial amount of the acetic acid liquid from the TPA solids. Residual catalyst is usually
20 separated from the slurry or cake terephthalic acid product by washing

(rinsing) the wet cake with catalyst-free acetic acid, water or other solvent.

The TPA solid is isolated by drying.

In the present invention, a novel process has been discovered resulting in fewer steps than the currently employed processes. The primary 5 utility of the invention is reduction of capital and operating costs associated with the isolation and drying of a terephthalic acid powder. In the conventional approach toward producing terephthalic acid via catalyzed oxidation of p-xylene in an acetic acid solvent, a slurry or cake terephthalic acid product is filtered, washed, then dried to produce a terephthalic acid 10 powder suitable as starting material for PET production.

In one embodiment of the present invention, the slurry or cake terephthalic acid product is filtered to produce a terephthalic acid cake with solvent and a solvent mother liquor stream. The terephthalic acid cake with solvent is then washed (rinsed) with water to recover residual metal catalyst 15 material and to produce a water-wet terephthalic acid cake and a solvent/water by-product liquor. The water-wet terephthalic acid cake is then combined with a diol to produce a terephthalic acid/diol mixture suitable as starting material in a commercial PET process. By eliminating conventional processes for isolating and drying a terephthalic acid solid, the 20 equipment and energy necessary to produce a terephthalic acid powder is also eliminated.

Another surprising and seemingly contradictory aspect of the invention is the benefit of addition of water to the acetic acid and ethylene glycol solvents. In general, in conventional processes for producing terephthalic acid, it is necessary to remove water produced in the oxidation process. Typically, use of acetic acid as an oxidation solvent necessitates an additional process step where acetic acid and water are separated. It is seemingly contradictory to produce an acetic acid and water mixture when it can be avoided by drying the terephthalic acid from the acetic acid solvent.

Additionally, in processes for producing PET via esterification of TPA with ethylene glycol, water is generated as a reaction by-product. In general, it is necessary to remove the water produced in the esterification process via an additional process step where ethylene glycol and water are separated. It is seemingly contradictory to produce an ethylene glycol and water mixture when it can be avoided by not introducing water with the TPA. However, a benefit of this invention is based on the premise that ethylene glycol/water and acetic acid/water separation systems normally exist for conventional TPA and PET production processes. In this invention, the value associated with eliminating the TPA drying can be of great benefit when compared to traditional TPA production processes.

SUMMARY OF THE INVENTION

The present invention relates to a process by which a carboxylic acid/diol mixture is obtained from a slurry or cake carboxylic acid product without isolation of a substantially dry carboxylic acid solid. More specifically, the present invention relates to a process for the production of a terephthalic acid/ethylene glycol mixture suitable as feedstock for the production of commercial PET. The resulting process can save energy and has fewer steps than currently employed processes. Specifically, the present invention incorporates a direct displacement of water with ethylene glycol. Incorporation of the displacement step eliminates the need to isolate a purified terephthalic acid solid and could eliminate the need for crystallization, solid-liquid separation, drying and solids handling equipment normally found in commercial purified terephthalic acid processes.

It is an object of this invention to provide a process for producing a carboxylic acid/diol mixture from a slurry or cake carboxylic acid product without isolation of a substantially dry carboxylic acid solid.

It is an object of this invention to provide a process for producing a carboxylic acid/diol mixture from a slurry or cake carboxylic acid product suitable as starting material for the production of polyesters or co-polyesters without isolation of a substantially dry carboxylic acid solid.

It is another object of this invention to provide a process for producing a terephthalic acid/diol mixture from a slurry or cake terephthalic acid product without isolation of a substantially dry terephthalic acid solid.

It is another object of this invention to provide a process for 5 producing a terephthalic acid/ethylene glycol mixture from a terephthalic acid solvent slurry or cake without isolation of a substantially dry terephthalic acid solid.

It is another object of this invention to provide a process for producing a terephthalic acid/ethylene glycol mixture without isolation of a 10 substantially dry terephthalic acid solid by removing water from a water-wet terephthalic acid cake through the use of a carboxylic acid/diol mixing zone.

In a first embodiment of this invention, a process for producing a carboxylic acid/diol mixture is provided, the process comprises:

- (a) removing in a liquor exchange zone impurities from a 15 carboxylic acid slurry to form a water-wet carboxylic acid cake, a mother liquor stream, a solvent mother liquor stream, and a solvent/water byproduct liquor stream;
- (b) routing the water-wet carboxylic acid cake to a vapor seal zone; and
- 20 (c) adding at least one diol to the water-wet carboxylic acid cake in a carboxylic acid/diol mixing zone to remove a portion of the water to form the carboxylic acid/diol mixture.

In another embodiment of this invention, a process for producing a carboxylic acid/diol mixture is provided, the process comprises:

- (a) removing in a solvent liquor exchange zone impurities from a carboxylic acid slurry to form a carboxylic acid cake with solvent, a mother liquor stream, and a solvent mother liquor stream;
- 5 (b) adding water in a water wash zone to the carboxylic acid cake with solvent to produce a water-wet carboxylic acid cake and a solvent/water by product liquor stream;
- (c) routing the water-wet carboxylic acid cake to a vapor seal 10 zone; and
- (d) adding at least one diol to the water-wet carboxylic acid cake in a carboxylic acid/diol mixing zone to remove a portion of the water to form the carboxylic acid/diol mixture.

In another embodiment of this invention, a process for producing a 15 carboxylic acid/diol mixture is provided, the process comprises:

- (a) removing in a solid-liquid separation zone impurities from a carboxylic acid slurry to form a slurry or cake carboxylic acid product and a mother liquor stream;
- 20 (b) removing in a solvent-water liquor exchange zone impurities from the slurry or cake carboxylic acid product to form a water-wet carboxylic acid cake, a solvent mother liquor stream, and a solvent/water byproduct liquor stream;

- (c) routing the water-wet carboxylic acid cake to a vapor seal zone; and
- (d) adding at least one diol to the water-wet carboxylic acid cake in a carboxylic acid/diol mixing zone to remove a portion of the water to 5 form the carboxylic acid/diol mixture.

In another embodiment of this invention, a process for producing a carboxylic acid/diol mixture is provided, the process comprises:

- (a) removing a solvent from a slurry or cake carboxylic acid product in a solvent-water liquor exchange zone; wherein a substantial 10 portion of the solvent in the slurry or cake carboxylic acid product is replaced with water to form a water-wet carboxylic acid cake;
- (b) routing the water-wet carboxylic acid cake to a vapor seal zone; and
- (c) adding at least one diol to the water-wet carboxylic acid cake 15 in a carboxylic acid/diol mixing zone to remove a portion of the water to form the carboxylic acid/diol mixture.

In another embodiment of this invention, a process for producing a terephthalic acid/diol mixture is provided, the process comprises:

- (a) removing in a solvent wash zone impurities from a slurry or 20 cake terephthalic acid product to form a terephthalic acid cake with acetic acid;

- (b) removing a substantial portion of a solvent in a water wash zone from the terephthalic acid cake with acetic acid to form a water-wet terephthalic acid cake; and
- (c) routing the water-wet terephthalic acid cake to a vapor seal zone; and
- (d) adding at least one diol to the water-wet terephthalic acid cake in a carboxylic acid/diol mixing zone to remove a portion of the water to form the terephthalic acid/diol mixture.

10 In another embodiment of this invention, a process for producing a terephthalic acid/diol mixture is provided, the process comprises:

- (a) removing a solvent from a slurry or cake terephthalic acid product in a solvent liquor exchange zone; wherein a substantial portion of the solvent in the slurry or cake terephthalic acid product is replaced with water to form a water-wet terephthalic acid cake;
- (b) routing the water-wet terephthalic acid cake to a vapor seal zone; and
- (c) adding at least one diol to the water-wet terephthalic acid cake in a carboxylic acid/diol mixing zone to remove a portion of the water to form the terephthalic acid/diol mixture.

20 In another embodiment of this invention, a process for producing a terephthalic acid/diol mixture is provided, the process comprises:

- (a) removing in a solvent wash zone impurities from a slurry or cake terephthalic acid product from a terephthalic acid cake with acetic

acid; wherein the solvent wash zone comprises at least one solid-liquid separation device that is operated at a temperature between about 40 °C to about 155 °C ;

- (b) removing a substantial portion of a solvent in a water wash zone from the terephthalic acid cake with acetic acid to form a water-wet terephthalic acid cake; wherein the water wash zone comprises at least one solid-liquid separation device that is operated at a temperature between about 40 °C to about 155 °C;
- (c) adding at least one diol to the water-wet terephthalic acid cake in a carboxylic acid/diol mixing zone to remove a portion of the water to form the terephthalic acid/diol mixture; wherein the adding occurs at a temperature between about 40 °C to about 290 °C; wherein the diol is ethylene glycol.

In another embodiment of this invention, a process for producing a carboxylic acid/diol mixture is provided, the process comprises:

- (a) removing in a solid-liquid separation zone impurities from a carboxylic acid slurry to form a slurry or cake carboxylic acid product and a mother liquor stream;
- (b) adding solvent to a slurry or cake carboxylic acid product in a solvent wash zone to the slurry or cake carboxylic acid product to produce a carboxylic acid cake with solvent and a solvent mother liquor stream;

- (c) adding water in a water wash zone to the carboxylic cake with solvent to produce a water-wet carboxylic acid cake and a solvent/water by product liquor stream;
- (d) routing the water-wet carboxylic acid cake to a vapor seal zone; and
- (e) adding at least one diol to the water-wet carboxylic acid cake in a carboxylic acid/diol mixing zone to remove a portion of the water to form the carboxylic acid/diol mixture.

In another embodiment of this invention, a process for producing a terephthalic acid/diol mixture is provided, the process comprises:

- (a) removing in a solid-liquid separation zone impurities from a crude terephthalic acid slurry to form a slurry or cake terephthalic acid product and a mother liquor stream;
- (b) adding solvent in a solvent wash zone to the slurry or cake terephthalic acid product to produce a terephthalic acid cake with solvent and a solvent mother liquor stream;
- (c) adding water in a water wash zone to the terephthalic acid cake with solvent to produce a water-wet terephthalic acid cake and a solvent/water by product liquor stream;
- (d) routing the water-wet terephthalic acid cake to a vapor seal zone; and

(e) adding at least one diol to the water-wet terephthalic acid cake in a carboxylic acid/diol mixing zone to remove a portion of the water to form the terephthalic acid/diol mixture.

These objects, and other objects, will become more apparent to
5 others with ordinary skill in the art after reading this disclosure.

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 illustrates one embodiment of this invention, a process for
10 producing a carboxylic acid/diol mixture.

Figure 2 illustrates another embodiment of this invention, a process for producing a carboxylic acid/diol mixture by utilizing a liquor exchange zone.

Figure 3 illustrates another embodiment of this invention, a process
15 for producing a carboxylic acid/diol mixture by utilizing a solvent-water liquor exchange zone.

Figure 4 illustrates another embodiment of this invention, a process for producing a carboxylic acid/diol mixture by utilizing a solvent liquor exchange zone.

In an embodiment of this invention shown in Figure 1, a process for producing a carboxylic acid/diol mixture **200** is provided. The process comprises:

Step (a) comprises optionally removing impurities from a carboxylic acid slurry **30** in an solid-liquid displacement zone **40** to form a slurry or cake carboxylic acid product **70** and a mother liquor stream **60**;

The carboxylic acid slurry comprises **30** at least one carboxylic acid, catalyst, at least one solvent, and impurities. The impurities typically comprise at least one or more of the following compounds: 4-
10 carboxybenzaldehyde(4-CBA), trimellitic acid(TMA), and 2,6-dicarboxyfluorenone(2,6-DCF). Suitable solvents include, but are not limited to, aliphatic mono-carboxylic acids, preferably containing 2 to 6 carbon atoms, or benzoic acid and mixtures thereof and mixtures of these compounds with water. Preferably the solvent is acetic acid mixed with
15 water, in a ratio of about 5:1 to about 99:1, preferably between about 8:1 and about 49:1. Throughout the specification acetic acid will be referred to as the solvent. However, it should be appreciated that other suitable solvents, such as those disclosed previously, may also be utilized. The solvent typically comprises acetic acid, but can be any solvent that has
20 been previously mentioned.

The carboxylic acid slurry **30** can be produced by oxidizing in a oxidation zone an aromatic feed stock. In one embodiment, the aromatic feedstock comprises paraxylene. The oxidation zone comprises at least one

oxidation reactor, and the carboxylic acid slurry comprises at least one carboxylic acid. The oxidation reactor can be operated at temperatures between about 120°C and about 250°C, preferably about 140°C to about 170°C. Typically the aromatic feed stock comprises paraxylene and the 5 carboxylic acid comprises terephthalic acid. In one embodiment of the invention the oxidation zone comprises a bubble column.

Therefore, for example, when terephthalic acid is utilized, the carboxylic acid slurry **30** would be referred to as terephthalic acid slurry and the carboxylic acid/diol mixture **200** would be referred to as a terephthalic 10 acid/diol mixture.

Carboxylic acids include any carboxylic acid produced via controlled oxidation of an organic precursor compound. For example carboxylic acids include aromatic dicarboxylic acids preferably having 8 to 14 carbon atoms, aliphatic dicarboxylic acids preferably having 4 to 12 carbon atoms, or 15 cycloaliphatic dicarboxylic acids preferably having 8 to 12 carbon atoms. Other examples of suitable carboxylic acids include, but are not limited to, terephthalic acid, benzoic, p-toulidic, isophthalic acid, trimellitic acid, naphthalene dicarboxylic acid, cyclohexanedicarboxylic acid, cyclohexanediacetic acid, diphenyl-4,4'-dicarboxylic acid, diphenyl-3,4'- 20 dicarboxylic acid, 2,2,-dimethyl-1,3-propandiol dicarboxylic acid, succinic acid, glutaric acid, adipic acid, azelaic acid, sebacic acid, and mixtures thereof.

Terephthalic acid slurry is conventionally synthesized via the liquid phase oxidation of paraxylene in the presence of suitable oxidation catalyst. Suitable catalysts include, but are not limited to, cobalt, manganese and bromine compounds, which are soluble in the selected solvent. In one embodiment of the invention the catalyst comprises cobalt, bromine and manganese. The cobalt and manganese combined can be in concentrations of about 100 ppm to about 2700 ppm by weight in the liquor. The bromine can be in concentrations of about 1000 ppm to about 2500 ppm by weight in the liquor.

10 The carboxylic acid slurry **30** is fed to a solid-liquid displacement zone **40** capable of removing a portion of the liquid contained in the carboxylic acid slurry **30** to produce a slurry or cake carboxylic acid product in conduit **70**. The removal of a portion of the liquid to produce a slurry or cake carboxylic acid product in conduit **70** can be accomplished by any means known in the art. A portion means at least 5% by weight of the liquid is removed. Typically, the solid-liquid displacement zone **40** comprises a solid-liquid separator that is selected from the group consisting of a decanter centrifuge, rotary disk centrifuge, belt filter, rotary vacuum filter, and the like. The carboxylic acid slurry in conduit **30** is fed to the solid-liquid displacement zone **40** comprising at least one solid-liquid separator.

15 The solid-liquid separator(s) can be operated at temperatures between about 50°C to about 200°C, preferably 140°C to about 170°C. The solid-liquid separator(s) can be operated at pressures between about 0 psig to

about 200 psig. The solid-liquid separator in the solid-liquid displacement zone **40** may be operated in continuous or batch mode, although it will be appreciated that for commercial processes, the continuous mode is preferred.

5 The impurities are displaced from the solid-liquid displacement zone **40** into a mother liquor stream and withdrawn via line **60**. In one embodiment of the invention, additional solvent is fed to the solid-liquid displacement zone **40** via line **50** to reslurry the carboxylic acid slurry **30** and form a slurry or cake carboxylic acid product **70**. When a terephthalic acid slurry is utilized in the solid-liquid separation zone **40**, a slurry or cake terephthalic acid product is produced. The slurry or cake terephthalic acid product typically comprises terephthalic acid and acetic acid. The mother liquor **60** is withdrawn from solid-liquid displacement zone **40** via line **60** and comprises a solvent, typically acetic acid, catalyst, and bromine compounds. The mother liquor in line **60** may either be sent to a process for separating impurities from oxidation solvent via lines not shown or recycled to the catalyst system via lines not shown. One technique for impurity removal from the mother liquor **60** commonly used in the chemical processing industry is to draw out or "purge" some portion of the recycle stream. Typically, the purge stream is simply disposed of or, if economically justified, subjected to various treatments to remove undesired impurities while recovering valuable components. Examples of impurity removal

10

15

20

processes include U.S. Patent 4,939,297 and U.S. Patent 4,356,319, herein incorporated by reference.

Step (b) comprises removing in a solvent wash zone **80** residual impurities from a slurry or cake carboxylic acid product **70** to form a 5 carboxylic acid cake with solvent **110** and a solvent mother liquor stream **100**.

Conduit **70** contains a slurry or cake carboxylic acid product **70** comprising a carboxylic acid, residual impurities and a solvent. The residual impurities comprise residual catalyst (typically but not limited to cobalt, manganese, or 10 bromine). Suitable solvents include, but are not limited to, aliphatic mono-carboxylic acids, preferably containing 2 to 6 carbon atoms, or benzoic acid and mixtures thereof and mixtures of these compounds with water.

Preferably, the solvent is comprised of mainly acetic acid and/or some water. The ratio of acetic acid to water can range from 50:50 to 98:2 acetic 15 acid to water by mass, more preferably in the range of 85:15 to 95:5, and most preferably in the range of 90:10 to 97:3. Suitable carboxylic acids include by are not limited to terephthalic acid, isophthalic acid, naphthalene dicarboxylic acid, trimellitic acid, and mixtures thereof.

The slurry or cake carboxylic acid product **70** is in the range of 10-20 90% by weight carboxylic acid. Preferably the slurry or cake carboxylic acid product **70** is in the range of 25- 40% by weight carboxylic acid for a slurry and in the range of 70-90% by weight for the cake product. Most preferably, the slurry or cake carboxylic acid product **70** is in the range of

30-40% by weight carboxylic acid. The slurry or cake carboxylic acid product in conduit **70** is then introduced into a solvent wash zone **80**, wherein a substantial portion of solvent is recovered in the solvent mother liquor stream in conduit **100**. The solvent mother liquor **102** comprises a substantial portion of the solvent. In one embodiment of the invention, additional solvent can be added via conduit **90** counter current to the flow of the slurry or cake carboxylic acid product **70** in the solvent wash zone **80**. The amount of stages of solvent counter current wash can be any amount of stages necessary to produce the carboxylic cake with solvent to the desired purity. Typically, the amount of stages in the solvent counter current wash can be about 1 to about 8, preferably about 2 to about 6, most preferably about 2 to about 4. For wash with more than one stage, counter current flow is preferable. Solvent counter current wash is preferable because typically it results in less solvent being used as compared to a process when solvent counter current wash is not utilized.

The solvent wash zone **80** comprises at least one solid-liquid separation device capable of efficiently separating solids and liquids. The solid-liquid separation device can typically be comprised of, but not limited to, the following types of devices: centrifuges, cyclones, rotary drum filters, belt filters, press filters, etc. The solvent wash zone **80** comprises at least one solid-liquid separation device(s) **110** which can operate within a temperature range of from approximately 40°C to 155°C. Preferably the solid-liquid separation device(s) **110** can operate within a temperature

range of from about 80°C to about 150°C. Most preferably the solid-liquid separation device(s) **110** can operate within a temperature range of from about 90°C to about 150°C. A carboxylic acid cake with solvent **110**, is produced wherein the moisture composition of the carboxylic acid cake with solvent **110** can be in the range of 0.5-30% by weight moisture, preferably in the range of 1-20% moisture, most preferably in the range of 1-10% moisture. Optionally, the residual solvent can be removed by a gas displacement step to minimize solvent contamination with wash. When the carboxylic acid is terephthalic acid and the solvent is acetic acid a terephthalic acid cake with acetic acid is produced.

Step (c) comprises optionally removing a substantial portion of a solvent in a water wash zone **120** from the carboxylic acid cake with solvent **110** to form a water-wet carboxylic acid cake **100** and a solvent/water byproduct liquor stream **140**.

The carboxylic acid cake with solvent **110**, is then subjected to a wash or "rinsing" with water or substantially water with residual amounts of solvent in the water wash zone **120**, wherein a substantial portion of the solvent is replaced with water to form a water-wet carboxylic acid cake **150**. The water-wet carboxylic acid cake **150**, is preferably in the range of about 0.5% to about 30% moisture, more preferably in the range of about 1 to about 20% moisture, and most preferably in the range of about 1% to about 10% moisture. The residual moisture of the water-wet carboxylic acid cake **150**, should contain less than about 2% solvent on a mass basis.

Additionally, the water-wet carboxylic acid cake **150** should contain less than 1% of any metals, preferably less than 100 ppm by weight, most preferably less than 10 ppm by weight, typically used as catalysts in p-xylene oxidation, in the slurry or cake carboxylic acid product in conduit **70**, 5 should remain in the water-wet carboxylic acid cake **150**. Examples of metals include but are not limited to cobalt, and manganese.

Wash water is introduced into the water wash zone **120** via conduit **130**. The wash water should be, on a continuous basis, comprise a mass feed rate in ratio with the solids in the carboxylic cake with solvent **110** in 10 the range of about 0.1:1 to about 1.5:1, preferably in the range of about 0.1:1 to about 0.6:1, most preferably in the range of about 0.2:1 to about 0.4:1. There are no limitations on the temperature or pressure of the wash water including the use of vaporized water, steam, or a combination of water and steam, as wash. In one embodiment of the invention, wash 15 water is introduced counter current to the carboxylic acid cake with solvent.

Additional wash water can be added via conduit **130** counter current to the flow of the carboxylic acid cake with solvent **110** in the water wash zone **120**. The amount of stages of water counter current wash can be any amount of stages necessary to produce the water wet carboxylic acid cake 20 to the desired purity. Typically, the amount of stages in the water counter current wash can be about 1 to about 8, preferably about 2 to about 6, most preferably about 2 to about 4. For wash with more than one stage, counter current flow is preferable. Water counter current wash is preferable

because typically it results in less water being used as compared to a process when water counter current wash is not utilized.

The water wash zone comprises a solid-liquid separation device **120** can typically be comprised of, but not limited to, the following types of devices: centrifuges, cyclones, rotary drum filters, belt filters, press filters, etc. The solid-liquid separation device can be operated within a temperature range of from about 40°C to about 155°C. Preferably, the second solid-liquid separation device can operate within a temperature range of from about 80°C to about 150°C. Most preferably, the second solid-liquid separation device can operate within a temperature range of from about 90°C to about 150°C

10

Optionally, the solvent/water byproduct liquor from the water wash zone **120**, is segregated from the solvent mother liquor stream produce by the solvent wash zone **80**.

15 Step (d) comprises routing the water-wet carboxylic acid cake **150** to a vapor seal zone **160**.

The water-wet carboxylic acid cake **150** is passed through a vapor seal zone **160** comprising a vapor seal device, and exits the vapor seal device via conduit **170**. The vapor seal device allows the water-wet carboxylic acid cake **150** to exit the counter current wash zone **120** but prevents diol from the carboxylic acid/diol mixing zone **180** from entering the counter current wash zone or any process zone proceeding the vapor seal zone **160**. The vapor seal device can be any device known in the art.

20

Examples include, but are not limited to rotary air lock valve, and solid conveying extruders.

Step (e) comprises adding at least one diol **190** to the water-wet carboxylic acid cake **170** in a carboxylic acid/diol mixing zone **180** to
5 remove a portion from the water-wet carboxylic acid cake **170** of the water to form the carboxylic acid/diol mixture **200**.

Finally, the water-wet carboxylic acid cake **170**, which is now substantially free of solvent is combined with a diol **190** in a carboxylic acid mixing zone **180**, to form a carboxylic acid/diol mixture **200** suitable for PET
10 production and other polyesters or co-polyesters. There are no special limitations on the carboxylic acid/diol mixing zone **180** with the exception that it comprises a device that must provide intimate contact between the water-wet carboxylic acid cake **170**, and the diol **190** to produce a the carboxylic acid/diol mixture **200**. Examples of such devices include, but
15 are not limited to the following: an agitated vessel, static mixer, screw conveyor, PET esterification reactor(s), etc. A solid eductor could be used to introduce the water-wet carboxylic acid cake into the device. Nor is there any specific limitation on the temperature range at which the device can operate. However, it is preferable that the temperature of device does not
20 exceed approximately 280°C, temperatures normally found within PET esterification reactors.

At least one diol in conduit **190** can be introduced in such a manner as to optionally displace the water as the dominant slurring liquid. This

can be accomplished by introducing a diol via conduit **190** as a saturated liquid at a temperature which is sufficient to vaporize the water. In one embodiment of the invention, the diol in conduit **190** is introduced as a saturated or superheated vapor. The diol in conduit **190** is at least one

5 selected from the group consisting of ethylene glycol, diethylene glycol, n-butylene glycol, i-butylene glycol, n-propylene glycol, 1,4 butanediol, cyclohexanedimethanol, and mixtures thereof. Preferably, the diol in conduit **190** is ethylene glycol. Note that within the system shown in figure 1, a substantially dry carboxylic acid solid is not formed. The primary

10 advantage in not forming a carboxylic acid dry solid is the elimination of solids handling equipment. Examples of solids handling equipment include but are not limited to a dryer, convey systems, and silos.

In other embodiments of this invention step (a), step (b) and step (c) can be combined into one zone known as the liquor exchange zone **250** as

15 shown in figure 2. The liquor exchange zone **250** comprises at least one solid-liquid separation device capable of performing the combined function of the solid-liquid separation zone **40**, the solvent wash zone **80** and the water wash zone **120** as previously described. Step (b) and step (c) can also be combined into one zone known as the solvent-water liquor

20 exchange zone **260** as shown in Figure 3. Finally step (a) and step (b) can be combined into one zone known as the solvent liquor exchange zone **270** as show in Figure 4. In each of the above embodiments comprises at least one solid-liquid separation device capable of performing the functions of the

combined zones as previously described. Examples of devices that can be used in the liquor exchange zone **250**, or the solvent-water liquor exchange zone **260**, or the solvent liquor exchange zone **270** included but are not limited to, the following type of devices centrifuges, cyclones, filters, and

5 such or combination thereof.